Q.P. Code: 16HS604

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SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR (AUTONOMOUS)

B.Tech I Year II Semester Supplementary Examinations February-2022 **ENGINEERING CHEMISTRY**

(Common to AGE, CE, EEE & ME)

Time: 3 hours

Max. Marks: 60

(Answer all Five Units $5 \times 12 = 60$ Marks)

UNIT-I

Define fuel cell. Explain the construction, uses and disadvantages of Methanol-Oxygen fuel cell.

12M

OR

a Write a note on Galvanic cell with suitable examples?

6M

b Explain in detail about Lithium ion batteries.

6M

UNIT-II

Describe the Ion exchange process for demineralization of water. What are its advantages and disadvantages? Give its regeneration process.

Describe the permutit process for softening of water. What are the advantages and 12M disadvantages of permutit process?

UNIT-III

a Write a note on synthetic petrol by Fischer-Tropsch process. 5

6M

b Write a short note on Aniline point & Neutralization number.

6M

OR

Describe the method employed for the refining of petroleum with a neat sketch.

12M

UNIT-IV

a Explain the preparation and uses of Bakelite.

6M

b What is polymer? Discuss the Preparation, Properties and uses of Teflon.

6M

OR

a Distinguish between Thermoplastics and thermosetting plastics.

6M

b Explain the procedures used in the processing of natural rubber.

6M

UNIT-V

a Explain in detail about setting and hardening of Portland cement.

6M

b Write a short note on properties and applications of Fullerenes and carbon nanotubes?

6M

OR

10 What are Refractories? Explain the classification, properties and failures of 12M Refractories.

*** END ***